

Surface Chemistry


 SET Self Evaluation Test - 14

- In which of the following commercial processes a catalyst is not used [CPMT 1989]
 - Haber's process
 - Deacon's process
 - Solvay process
 - Lead chamber process
- A catalyst [IIT 1984; AFMC 1995; CBSE PMT 1995]
 - Increases the average kinetic energy of reacting molecules
 - Increases the activation energy
 - Alters the reaction mechanism
 - Increases the frequency of collisions of reacting species
- The coagulation of 100 ml of a colloidal sol of gold is completely prevented by addition of 0.25 g of a substance 'X' to it before adding 10 ml of 1% NaCl solution. The gold number of 'X' is
 - 0.25
 - 25
 - 250
 - 2.5
- Which one of the following is not a property of hydrophilic sols
 - High concentrations of dispersed phase can be easily attained
 - Coagulation is reversible
 - Viscosity and surface tension are about the same as for water
 - The charge of the particle depends on the pH values of the medium; it may be positive, negative or even zero
- Peptising agent is
 - Always an electrolyte
 - Always a non-electrolyte
 - Electrolyte or non-electrolyte
 - A lyophilic colloid
- The catalyst used in the manufacture of methanol from water gas is [MP PET 2002]
 - V_2O_5
 - $Ni + Mo$
 - $ZnO + Cr_2O_3$
 - $Pt + W$
- Organic catalysts differ from inorganic catalysts [AFMC 1989]
 - By acting at very high temperature
 - By acting at low temperature
 - Being used up
 - Being proteinous in nature
- Commercial detergents mainly contain [CPMT 1993]
 - $RCOONa$
 - $RONa$
 - $RSNa$
 - $ROSO_2Na$
- In which of the following processes shape-selective catalysis is occurring
 - Conversion of alcohol to gasoline
 - Synthesis of methanol from CO and H_2
 - Polymerisation of ethylene
 - Synthesis of ammonia
- Which one of the following is used for reviving the exhausted permutite [EAMCET 2003]
 - HCl solution
 - 10% $CaCl_2$ solution
 - 10% $MgCl_2$ solution
 - 10% NaCl solution
- The ability of a catalyst to accelerate the chemical reaction is known as [CPMT 2000; KCET 2000]
 - Selectively
 - Activity
 - Negative catalyst
 - None of these
- $AlCl_3$, in reactions, acts as
 - Oxidizing agent
 - Reducing agent
 - Acid catalyst
 - None of these
- On addition of one ml of 10% NaCl solution to 10 ml gold sol in the presence of 0.25 gm of starch. The coagulation is just prevented, starch has gold number [MP PET 2004]
 - 0.025
 - 0.25
 - 2.5
 - None
- Milk is an example of [MP PET 2001; JIPMER 2002; MP PMT 2002,04]
 - Pure solution



(b) Gel

(c) Emulsion

(d) Suspension

AS Answers and Solutions

(SET -14)

- (c) In Haber's process we use iron as a catalyst, In Deacon's process we use Cu_2Cl_2 as a catalyst and in lead chamber process we use N_2O_5 as a catalyst but in solvay process no catalyst is used.
- (d) A catalyst can increase the rate of reaction and hence increases the frequency of collision of reacting species.
- (b) 250 mg of X is present in 100 ml of colloidal sol of gold. By definition, gold no. of X is that amount of it in mg which is present in 10 ml of colloidal gold solution. Hence in 10 ml, the amount of X present is 25 mg which is the gold number of X.
- (c) Viscosity and surface tension are not same for water in hydrophilic sols.
- (a) Process of converting precipitate into colloidal particles by adding suitable electrolyte called peptisation and stabilizing agent (electrolytes) as peptizing agent.
- (c) $[CO + H_2] + H_2 \xrightarrow{ZnO + Cr_2O_3} CH_3OH$
- (d) Organic catalyst are proteinous in nature and obtain from living cell.
- (a) Commercial detergents mainly contain salts of higher fatty acids.
- (a,c) Shape selective catalyst are zeolites and zeolites are aluminosilicates of general formula $M_{x/n}[(AlO_2)_x \cdot (SiO_2)_y] \cdot mH_2O$. Zeolites are used in conversion of alcohol to gasoline and in polymerisation of ethylene.
- (d) 10% $NaCl$ solution is used for reviving the exhaust permutite.
- (b) The ability of a catalyst to accelerate the chemical reaction is known as its activity. Degree of acceleration can be as high as 10^{10} times in certain reactions.
- (c) $AlCl_3$ may be written as $Al^{+3}Cl^-$ and all the tri-positive ions are Lewis acid hence it can be acts as acid catalyst.
- (d) $0.25 \times 1000 = 250$
- (c) Milk is the example of emulsion of fat globules in water.

